

DO NOT WRITE ON THIS CARD

Relative Atomic Masses

based on $^{12}\text{C} = 12.0000$

	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
1	1 H Hydrogen 1.0079																	2 He Helium 4.0026
2	3 Li Lithium 6.941	4 Be Beryllium 9.0122											5 B Boron 10.81	6 C Carbon 12.011	7 N Nitrogen 14.007	8 O Oxygen 15.999	9 F Fluorine 18.998	10 Ne Neon 20.180
3	11 Na Sodium 22.990	12 Mg Magnesium 24.305											13 Al Aluminium 26.982	14 Si Silicon 28.086	15 P Phosphorus 30.974	16 S Sulfur 32.06	17 Cl Chlorine 35.453	18 Ar Argon 39.948
4	19 K Potassium 39.098	20 Ca Calcium 40.08	21 Sc Scandium 44.956	22 Ti Titanium 47.87	23 V Vanadium 50.942	24 Cr Chromium 51.996	25 Mn Manganese 54.938	26 Fe Iron 55.845	27 Co Cobalt 58.933	28 Ni Nickel 58.69	29 Cu Copper 63.546	30 Zn Zinc 65.41	31 Ga Gallium 69.72	32 Ge Germanium 72.64	33 As Arsenic 74.922	34 Se Selenium 78.96	35 Br Bromine 79.904	36 Kr Krypton 83.80
5	37 Rb Rubidium 85.468	38 Sr Strontium 87.62	39 Y Yttrium 88.906	40 Zr Zirconium 91.224	41 Nb Niobium 92.906	42 Mo Molybdenum 95.94	43 Tc Technetium (98)	44 Ru Ruthenium 101.07	45 Rh Rhodium 102.91	46 Pd Palladium 106.42	47 Ag Silver 107.87	48 Cd Cadmium 112.41	49 In Indium 114.82	50 Sn Tin 118.71	51 Sb Antimony 121.76	52 Te Tellurium 127.60	53 I Iodine 126.904	54 Xe Xenon 131.29
6	55 Cs Caesium 132.91	56 Ba Barium 137.33	*	72 Hf Hafnium 178.49	73 Ta Tantalum 180.95	74 W Tungsten 183.84	75 Re Rhenium 186.21	76 Os Osmium 190.2	77 Ir Iridium 192.22	78 Pt Platinum 195.08	79 Au Gold 196.97	80 Hg Mercury 200.59	81 Tl Thallium 204.38	82 Pb Lead 207.2	83 Bi Bismuth 208.98	84 Po Polonium (209)	85 At Astatine (210)	86 Rn Radon (222)
7	87 Fr Francium (223)	88 Ra Radium (226)	**	104 Rf Rutherfordium (261)	105 Db Dubnium (262)	106 Sg Seaborgium (266)	107 Bh Bohrium (264)	108 Hs Hassium (277)	109 Mt Meitnerium (268)	110 Ds Darmstadtium (271)	111 Rg Roentgenium (272)	112 Uub Ununbium (285)	113 Uut Ununtrium (284)	114 Uuq Ununquadium (289)	115 Uup Ununpentium (288)	116 Uuh Ununhexium (293)	117 Uus Ununseptium	118 Uuo Ununoctium (294)

*Lanthanoids	57 La Lanthanum 138.905	58 Ce Cerium 140.12	59 Pr Praseodymium 140.91	60 Nd Neodymium 144.24	61 Pm Promethium (145)	62 Sm Samarium 150.36	63 Eu Europium 151.96	64 Gd Gadolinium 157.25	65 Tb Terbium 158.93	66 Dy Dysprosium 162.50	67 Ho Holmium 164.93	68 Er Erbium 167.26	69 Tm Thulium 168.93	70 Yb Ytterbium 173.04	71 Lu Lutetium 174.97
**Actinoids	89 Ac Actinium (227)	90 Th Thorium 232.04	91 Pa Protactinium 231.04	92 U Uranium 238.03	93 Np Neptunium (237)	94 Pu Plutonium (244)	95 Am Americium (243)	96 Cm Curium (247)	97 Bk Berkelium (247)	98 Cf Californium (251)	99 Es Einsteinium (252)	100 Fm Fermium (257)	101 Md Mendelevium (258)	102 No Nobelium (259)	103 Lr Lawrencium (262)

CHEMICAL DATA CARD

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PHYSICAL CONSTANTS

Avogadro's constant	N_A or L	$6.022 \times 10^{23} \text{ mol}^{-1}$
Bohr radius	a_0	$5.292 \times 10^{-11} \text{ m}$
Boltzmann's constant	k	$1.381 \times 10^{-23} \text{ J K}^{-1}$
Electron rest mass	m_e	$9.110 \times 10^{-31} \text{ kg}$
Electronic charge	e	$1.602 \times 10^{-19} \text{ C}$
Faraday	F	$9.649 \times 10^4 \text{ C mol}^{-1}$
Gas constant	R	$8.314 \text{ J K}^{-1} \text{ mol}^{-1}$
Molar volume, ideal gas, 0 °C, 1 bar	V_m^\ominus	$22.71 \text{ dm}^3 \text{ mol}^{-1}$
Nernst factor	2.303 RT/F	0.059 V at 25 °C
Neutron rest mass	m_n	$1.675 \times 10^{-27} \text{ kg}$
Permittivity of vacuum	ϵ_0	$8.854 \times 10^{-12} \text{ J}^{-1} \text{ C}^2 \text{ m}^{-1}$
	$4\pi\epsilon_0$	$1.113 \times 10^{-10} \text{ J}^{-1} \text{ C}^2 \text{ m}^{-1}$
Planck's constant	h	$6.626 \times 10^{-34} \text{ J s}$
Proton rest mass	m_p	$1.673 \times 10^{-27} \text{ kg}$
Rydberg's constant	R_H	$1.097 \times 10^7 \text{ m}^{-1}$
Triple point temperature of water	-	273.16 K
Temperature, conventional reference	T	298.15 K (25 °C)
Velocity of light	c	$2.998 \times 10^8 \text{ m s}^{-1}$

PREFIXES

f	p	n	μ	m	c	d	k	M	G	T
femto	pico	nano	micro	milli	centi	deci	kilo	mega	giga	tera
10^{-15}	10^{-12}	10^{-9}	10^{-6}	10^{-3}	10^{-2}	10^{-1}	10^3	10^6	10^9	10^{12}

CONVERSION FACTORS AND DEFINITIONS

Ångstrom	Å	10^{-10} m
Atmosphere	atm	$1.013 \times 10^5 \text{ Pa (or N m}^{-2}\text{)}$
		760 Torr
Atomic mass unit (a.m.u.)	m_u	$1.661 \times 10^{-27} \text{ kg}$
Bar	bar	$10^5 \text{ Pa (or N m}^{-2}\text{)}$
Bohr magneton	μ_B	$9.273 \times 10^{-24} \text{ A m}^2$
Calorie	cal	4.184 J
Coulomb	C	A s
Debye	D	$3.336 \times 10^{-30} \text{ C m}$
Electron volt	eV	$1.602 \times 10^{-19} \text{ J}$
		(or 96.485 kJ mol ⁻¹)
Hartree		$4.3598 \times 10^{-18} \text{ J}$
Frequency	Hz	s ⁻¹
Joule	J	N m (or kg m ² s ⁻² or V C)
Kelvin temperature	T/K	T/°C + 273.15
Litre	l	1.0 dm^3 (or 10^{-3} m^3)
Newton	N	J m^{-1} (or kg m s ⁻²)
Nuclear magneton	μ_N	$5.051 \times 10^{-27} \text{ A m}^2$
Pascal	Pa	N m^{-2} (or kg m ⁻¹ s ⁻²)
Torr (or mm of mercury)	Torr	$1.333 \times 10^2 \text{ Pa (or N m}^{-2}\text{)}$
		$1.333 \times 10^{-3} \text{ bar}$
Volt	V	J C^{-1} (or kg m ² A ⁻¹ s ⁻³)
Watt	W	J s^{-1} (or kg m ² s ⁻³)